

DROP TEST

P/M & P/T ALKALINITY (1 drop = 10 ppm)

COMPONENTS:

- | | |
|---------------|--|
| 1 x 5067G | Instruction |
| 1 x 9198G | Sample Tube, Graduated (25 mL) w/ cap & green dot, plastic |
| 1 x R-0637-C | Methyl Orange Indicator, 2 oz, DB |
| 1 x R-0638G-A | Phenolphthalein Indicator, .75 oz w/ green cap, DB |
| 1 x R-0645-C | Total Alkalinity Indicator, 2 oz, DB |
| 1 x R-0687G-C | Sulfuric Acid .12N, 2 oz w/ green cap, DB |

**TO ORDER REPLACEMENT PARTS AND REAGENTS CALL TOLL-FREE
800-TEST KIT (800-837-8548).**

PROCEDURE:

**CAREFULLY READ AND FOLLOW PRECAUTIONS ON REAGENT LABELS.
KEEP REAGENTS AWAY FROM CHILDREN.**

NOTE: When dispensing reagents from dropper bottles, **always** hold bottle in a vertical position.

P/M Alkalinity Test

1. Rinse and fill 25 mL sample tube (#9198G) to 25 mL mark with water to be tested.

NOTE: For results in grains per gallon (gpg), fill to 14.6 mL mark.

2. Add 3 drops R-0638G Phenolphthalein Indicator. Swirl to mix. Sample will turn pink (Fig. 1) if P alkalinity is present—proceed to Step 3. If sample is colorless, proceed to Step 4.
3. Add R-0687G Sulfuric Acid .12N dropwise, swirling and counting after each drop, until color just changes from pink to colorless. Record drops as P reading.

4. Add 5 drops R-0637 Methyl Orange Indicator. Swirl to mix. Sample will turn yellow (Fig. 2).

5. Add R-0687G Sulfuric Acid .12N dropwise, swirling and counting after each drop, until color changes from yellow to orange (salmon pink) (Fig. 3). Record total drops (Steps 3 & 5) as M reading.

6. Multiply P reading by 10. Record as parts per million (ppm) P alkalinity as calcium carbonate. Multiply M reading by 10. Record as ppm M alkalinity as calcium carbonate.

NOTE: For 14.6 mL sample, record P reading as grains per gallon (gpg) P alkalinity as calcium carbonate. Record T reading as gpg T alkalinity as calcium carbonate.

(OVER)



Fig. 1



Fig. 2



Fig. 3

DROP TEST

P/M & P/T ALKALINITY (1 drop = 10 ppm)

Instr. #5067G

P/T Alkalinity Test

1. Rinse and fill 25 mL sample tube (#9198G) to 25 mL mark with water to be tested.

NOTE: For results in grains per gallon (gpg), fill to 14.6 mL mark.

2. Add 3 drops R-0638G Phenolphthalein Indicator. Swirl to mix. Sample will turn pink (Fig. 4) if P alkalinity is present—proceed to Step 3. If sample is colorless, proceed to Step 4.

3. Add R-0687G Sulfuric Acid .12N dropwise, swirling and counting after each drop, until color just changes from pink to colorless. Record drops as P reading.

4. Add 5 drops R-0645 Total Alkalinity Indicator. Swirl to mix. Sample will turn green (Fig. 5).

5. Continue adding R-0687G Sulfuric Acid .12N dropwise, swirling and counting after each drop, until color changes from green to red (Fig. 6). Record total drops (Steps 3 & 5) as T reading.

6. Multiply P reading by 10. Record as parts per million (ppm) P alkalinity as calcium carbonate. Multiply T reading by 10. Record as ppm T alkalinity as calcium carbonate.

NOTE: For 14.6 mL sample, record P reading as grains per gallon (gpg) P alkalinity as calcium carbonate. Record T reading as gpg T alkalinity as calcium carbonate.



Fig. 4



Fig. 5



Fig. 6



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DROP TEST CHLORIDE (1 drop = 10 ppm)

Instr. #5067O

COMPONENTS:

- 1 x 5067O Instruction
- 1 x 9198O Sample Tube, Graduated (25 mL) w/ cap & orange dot, plastic
- 1 x R-0630-C Chromate Indicator, 2 oz, DB
- 1 x R-0638O-A Phenolphthalein Indicator, .75 oz w/ orange cap, DB
- 1 x R-0687O-C Sulfuric Acid .12N, 2 oz w/ orange cap, DB
- 1 x R-0706-C Silver Nitrate Reagent, 2 oz, DB

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NOTE: When dispensing reagents from dropper bottles, **always** hold bottle in a vertical position.

Chloride Test

NOTE: When sulfite content of sample water to be tested exceeds 10 ppm, the sulfite should be oxidized to prevent interference in test. A 25 mL water sample is first adjusted to the appropriate pH, then 1 mL (or 25 drops) of R-0649 Hydrogen Peroxide Solution (sold separately) is added and thoroughly mixed. Continue with the rest of the procedure.

1. Rinse and fill 25 mL sample tube (#9198O) to 25 mL mark with water to be tested.
2. Add 2 drops R-0638O Phenolphthalein Indicator. Swirl to mix. If sample turns pink (Fig. 1), add R-0687O Sulfuric Acid .12N dropwise, swirling after each drop, until color changes from pink to colorless.

3. Add 5 drops R-0630 Chromate Indicator. Swirl to mix. Sample will turn yellow (Fig. 2).

4. Add R-0706 Silver Nitrate Reagent dropwise, swirling and counting after each drop, until color changes from yellow to milky salmon (brick) red (Fig. 3).

NOTE: Do not add enough R-0706 Silver Nitrate Reagent to give a brown color. First change from yellow to a milky salmon (brick) red is the endpoint.

5. Multiply drops of R-0706 Silver Nitrate Reagent by 10. Record as parts per million (ppm) chloride.



Fig. 1



Fig. 2



Fig. 3



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DROP TEST

TOTAL HARDNESS (1 drop = 10 ppm)

Instr. #5067B

COMPONENTS:

| | |
|---------------|---|
| 1 x 5067B | Instruction |
| 1 x 9198B | Sample Tube, Graduated (25 mL) w/ cap & blue dot, plastic |
| 1 x R-0619B-C | Hardness Buffer, 2 oz w/ blue cap, DB |
| 1 x R-0620B-I | Hardness Indicator Powder, 10 g w/ blue dot |
| 1 x R-0683-C | Hardness Reagent, 2 oz, DB |

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PROCEDURE:

CAREFULLY READ AND FOLLOW PRECAUTIONS ON REAGENT LABELS.
KEEP REAGENTS AWAY FROM CHILDREN.

NOTE: When dispensing reagents from dropper bottles, **always** hold bottle in a vertical position.

Total Hardness Test

1. Rinse and fill 25 mL sample tube (#9198B) to 25 mL mark with water to be tested (Fig. 1).
2. Add 5 drops R-0619B Hardness Buffer. Swirl to mix.
3. Add 1 dipper R-0620B Hardness Indicator Powder. Swirl until dissolved. Sample will turn red (Fig. 2) if hardness is present.
4. Add R-0683 Hardness Reagent dropwise, swirling and counting after each drop, until color changes from red to blue (Fig. 3).

5. Multiply drops of R-0683 Hardness Reagent by 10. Record as parts per million (ppm) total hardness as calcium carbonate.



Fig. 1



Fig. 2



Fig. 3



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DROP TEST SODIUM SULFITE (1 drop = 10 ppm)

Instr. #5067W

COMPONENTS:

| | |
|---------------|--|
| 1 x 5067W | Instruction |
| 1 x 9198W | Sample Tube, Graduated (25 mL) w/ cap & white dot, plastic |
| 1 x R-0638W-C | Phenolphthalein Indicator, 2 oz w/ white cap, DB |
| 1 x R-0699-C | Iodide Iodate Reagent, 2 oz, DB |
| 1 x R-0725-I | Acid Starch Indicator Powder, 10 g |

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Sodium Sulfite Test

NOTE: Sample must be cooled to less than 100°F (38°C) to prevent high test results. Sample must be protected from air contact while cooling to prevent low test results.

1. Collect water to be tested in a clean, preferably large-mouthed, bottle to overflowing. Immediately cap and cool to room temperature.
2. Rinse and fill 25 mL sample tube (#9198W) to 25 mL mark with cooled (room temperature) water to be tested (Fig. 1).

NOTE: For results in grains per gallon (gpg), fill to 14.6 mL mark.

3. Add 1 drop R-0638W Phenolphthalein Indicator. Swirl to mix. Sample will turn pink (Fig. 2).

4. Add R-0725 Acid Starch Indicator Powder a dipper at a time, swirling after each dipper, until color changes from pink to colorless. Add 2 more dippers. Swirl until dissolved.

5. Add R-0699 Iodide Iodate Reagent dropwise, swirling and counting after each drop, until sample changes from colorless to a faint but permanent blue (Fig. 3).

6. Multiply drops of R-0699 Iodide Iodate Reagent by 10. Record as parts per million (ppm) sodium sulfite.

NOTE: For 14.6 mL sample, record drops as grains per gallon (gpg) sodium sulfite.

NOTE: For results as sulfite, multiply sulfite concentration by 0.64.



Fig. 1



Fig. 2

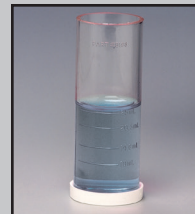


Fig. 3



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